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Synthetic studies of the HIV-1 protease inhibitive didemnaketals: stereocontrolled synthesis of an ester side chain $\stackrel{\approx}{\sim}$

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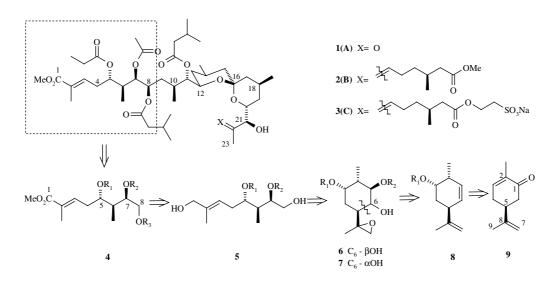
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Abstract—The stereocontrolled synthesis of the C_1 – C_8 portion, the ester side chain of the HIV-1 protease inhibitive didemnaketals from the ascidian *Didemnum* sp., has been carried out through 15 steps starting from (*S*)-carvone as the chiral template. This approach involved the diastereoselective construction of three conjoint chiral centers by intramolecular chiral inducement, and generation of allylic alcohol intermediate through a key Grob fragmentation reaction. © 2004 Elsevier Ltd. All rights reserved.

In 1991, Faulkner and co-workers reported the novel didemnaketals **A** (1) and **B** (2) (Scheme 1) isolated from the ascidian *Didemnum* sp. at Auluptagel Island, Palau, and demonstrated **A** and **B** exhibited to be highly inhibitive to HIV-1 protease with the IC₅₀ being 2 and $10 \,\mu$ M, respectively.¹ In-depth investigation proved the ester side chain might be the main activity portion of the HIV-1 protease didemnaketals.² As their important biological activity and particular structure, we have

already performed the synthesis of the spiroketal moiety.³ In 2002, Faulkner and co-workers finished the full assignment of relative and absolute stereochemistry of the didemnaketals,^{1c} which benefits our further studies on the asymmetric total synthesis of this kind of compounds. Here, we present our efficient and stereocontrolled synthetic of C_1-C_8 side-chain **4** of the didemnaketals, which have three conjoint stereocenters (C_5-C_7) and the conjugated unsaturated ester.



Scheme 1.

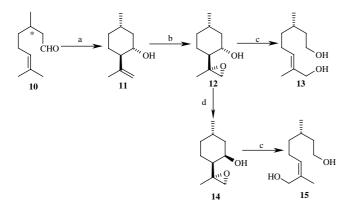
Keywords: Ester side chain; HIV-1 protease; Intramolecular chiral inducement; Grob fragmentation reaction.

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Our synthesis, shown in the retrosynthetic analysis (Scheme 1), was based on the selection of the menthane kind of monoterpene (*S*)-carvone **9** as starting material, for its C₁–carbonyl, C₂–Me, and the isopropene (C₇–C₉) could generate the corresponding C₇–OH, C₆–Me, and the terminal isopropyl ester of **4**, respectively. So a C₆– hydroxy epoxide intermediate **6** (or **7**) was designed and a key Grob fragmentation would be used to generate the open-chained allylic alcohol **5**.⁴

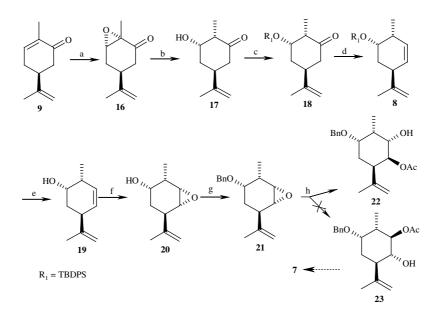
In the initial investigation on the possibility of above synthetic analysis, we tested the key Grob fragmentation using the model substrate **12** prepared from (\pm) -citronellal **10**.⁵ It was very interesting that we observed the configuration of the double bond formed in Grob fragmentation of the epoxide **12** was highly dependent upon the inducement of its OH group. As shown in Scheme 2, if fragmentation of the epoxide **12** bearing the



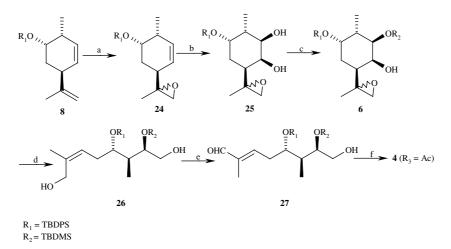
Scheme 2. Reagents and conditions: (a) ZnBr_2 , CH_2Cl_2 , 90%; (b) *m*-CPBA, CH₂Cl₂; 0 °C, 92%; (c) Al(^{*i*}PrO)₃, ^{*i*}PrOH, reflux, 90%; (d) (i) *p*-NO₂C₆H₄CO₂H, PPh₃, DEAD, benzene; (ii) K₂CO₃, MeOH, 72% (2 steps).

trans-OH and epoxyisopropyl located at the six-membered ring (prepared from 11 via direct *m*-CPBA epoxidation) was promoted with Al(ⁱPrO)₃, only (*E*)-8hydroxycitronellol 13 was produced.⁶ However, if the substrate 14 possessed the *cis*-OH and epoxyisopropyl (prepared from 12 via Mitsunobu reaction), a (*Z*)-8hydroxycitronellol 15 was obtained as a single product. This important fact provided alternative procedure for synthesis of the hard-to-available *Z*-double bond, and here suggested that the intermediates 7 could be a favorable precursor for constructing the side-chain 4 with *E*-double bond by use of Grob fragmentation.

On the basis of above model testing, the key intermediate 7 was designed firstly. The synthesis began with the stereoselective epoxidation of commercially available (S)-(+)-carvone (9) using basic hydrogen peroxide⁷ (Scheme 3). Successive organoselenium-mediated catalytic reductive ring opening of the resulting oxirane 16 led to the α -hydroxyl ketone 17.⁸ Therefore the corresponding C_5 -OH of 4 was stereoselectively introduced. As the direct protection of C₃–OH in 17 with TBDPSCl led to substantial elimination to give the starting carvone 9, protection of the C_3 -OH had to be performed and the product 18 was obtained in three steps: protecting the carbonyl group in 17 with ethylene glycol, then protecting the hydroxyl group with TBDPSCl and again releasing carbonyl group with TsOH. Subsequently elimination of the carbonyl of 18 by hydrazonization with TsNHNH₂ followed by treatment the formed hydrazone with *n*-BuLi afforded diene 8. With 8 in hand we turned our attention to gain the key intermediate 23. Thus deprotection of TBDPS of 8 with TBAF followed by stereoselective epoxidation of the formed homoallylic alcohol 19 with TBHP through the inducement of the hydroxyl group and then protection of the hydroxy in 20 afforded the epoxide 21. Here, the



Scheme 3. Reagnets and conditions: (a) H_2O_2 , NaOH, MeOH, 95%; (b) (PhSe)₂, *N*-acetylcysteine, NaOH, MeOH, 70%; (c) (i) (CH₂OH)₂, PPTS, benzene, reflux; (ii) TBDPSCI, NaH, THF, 0 °C; (iii) TsOH, acetone, 72% (three steps); (d) (i) TsNHNH₂, MeOH; (ii) *n*-BuLi, THF, -78 °C, rt, 52% (2 steps); (e) TBAF, THF, reflux, 84%; (f) *t*-BuO₂H, V(acac)₂, 89%; (g) BnBr, NaH, TBAI, THF, 0 °C, 68%; (h) HOAc, NaOAc, reflux, 72%.



Scheme 4. Reagents and conditions: (a) *m*-CPBA, CH₂Cl₂, 0 °C, 65%; (b) K_2OsO_4 , NMO, 'BuOH–acetone–H₂O (1:2:1) 75%; (c) TBSCl, imid., DMF, 75%; (d) Al('PrO)₃, 'PrOH, 72%; (e) SeO₂, EtOH (95%), reflux, 72%; (f) (i) Ac₂O, DMAP, Py; (ii) NaClO₂, 2-methyl-2-butene, NaH₂PO₄, 'BuOH/H₂O; (iii) CH₂N₂, Et₂O, 70% (three steps).

deprotection of TBDPS group in 8 was necessary for the regioselective monoepoxidation of 19 because a direct epoxidation of 8 without inducement of C_3 -OH only gave a terminal epoxy product. Unfortunately, the following ring opening of the epoxide 21 with NaOAc could not afford the desired homoallylic alcohol 23 (which could be easily converted to the key intermediate 7) but the alcohol 22, which had two opposite stereocenters C_1 and C_6 to 23 and was of no use to our synthesis. The possible reason for this result would be that the diaxial transition state for yielding 22 is more stable than the diequatorial one, which lead to the product 23.⁹

Subsequently we had to modify the above route for first demanding the desired C_7 stereochemistry of 4 $(R_3 = Ac)$. A viable route as shown in Scheme 4 was performed by regioselective epoxidation of the diene 8 with m-CPBA and the terminal epoxide 24 in two isomers (1:1) was obtained. Then the stereoselective osmylation¹⁰ of the epoxide 24 followed by selective protection of the equatorial hydroxyl group in the formed diol 25 afforded the intermediate 6, in which three desired stereocenters corresponding C_5-C_7 have been constructed successfully. Then fragmentation of epoxide alcohol 6 with $Al(^{i}PrO)_{3}$ afforded the Z-allylic alcohol 26, whose stereochemistry for the double bond and three stereocenters was confirmed through spectroscopy inspection of its derivatives and the configuration of the double bond was just what we anticipated before.¹¹ The formation of only one isomer of **26** from two isomers of 6 indicated that the configuration of epoxide has no effect on the Grob fragmentation reaction. Fortunately, the following selective oxidation of the Z-primary allylic alcohol 26 with selenium (IV) oxide afforded the E-unsaturated aldehyde 27 in 70% yield.¹² Protection of the hydroxyl group in the aldehyde 27 with acetic anhydride followed by oxidation of the formed aldehyde with sodium chlorite and then esterification of the formed acid with CH_2N_2 afforded the methyl ester 4 ($R_3 = Ac$) successively.¹³ Further total

synthetic studies on the didemnaketals are ongoing in our group.

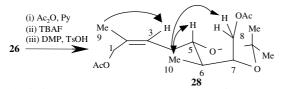
Acknowledgements

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- 11. The compound **26** was first made into derivative **28** through three steps, and the stereochemistry of **28** was confirmed through ¹H NMR NOE experiment and NO-ESY experiment as shown below. For example, irradiation of C₉–H (δ : 1.76 ppm) lead to 4% enhancement of C₃–H (δ : 5.48 ppm), irradiation of C₅–H (δ : 3.28 ppm) lead to 5% enhancement of C₁₀–H (δ : 0.86 ppm), and irradiation of

 C_{10} -H (δ : 0.86 ppm) lead to 4% enhancement of C_5 -H (δ : 3.28 ppm) and 4% enhancement of C_8 -H (δ : 4.05 ppm)



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- 13. Analytical data of the compound 4: ¹H NMR (300 MHz, CDCl₃): δ 7.66–7.62 (m, 4H), 7.44–7.32 (m, 6H), 6.71 (t, 1H, J = 6.7 Hz), 3.90–3.75 (m, 4H), 3.67 (s, 3H), 2.37–2.27 (m, 2H), 1.93 (s, 3H), 1.87–1.83 (m, 1H), 1.65 (s, 3H), 1.04 (s, 9H), 0.95 (d, 3H, J = 7.2 Hz), 0.82 (s, 9H), -0.01 (s, 3H), -0.08 (s, 3H); ¹³C NMR (75 MHz): δ 170.6, 168.3, 139.9, 136.0 (4C), 133.9, 133.8, 129.7 (2C), 128.4, 127.6 (2C), 127.5 (2C), 74.6, 71.7, 67.0, 51.5, 42.5, 32.9, 27.0 (3C), 25.8 (3C), 20.8, 19.3, 18.1, 12.5, 10.0, -4.3, -4.6; FAB-MS (M+H)⁺, m/z 627; HR-SIMS: m/z calcd for C₃₅H₃₅Si₂O₆ (M+H) 627.3532; found 627.3523.